## **Chemical Engineering Thermodynamics MCQ Questions**

- 1. First law of thermodynamics deals with the
  - a. Direction of energy transfer
  - b. Reversible processes only
  - c. Irreversible processes only
  - d. None of these

Ans: A

- 2. Second law of thermodynamics is concerned with the
  - a. Amount of energy transferred
  - b. Direction of energy transfer
  - c. Irreversible processes only
  - d. Non cyclic processes only Ans: B
- 3. Fugacity and pressure are numerically equal, when the gas is
  - a. in standard state
  - b. at high pressure
  - c. at low temperature
  - d. in ideal state

Ans: D

- 4. Ideal refrigeration cycle is
  - a. same as Carnot cycle
  - b. same as reverse Carnot cycle
  - c. dependent on the refrigerant's properties
  - d. the least efficient of all refrigeration processes
    - Ans: B
  - Which of the following is a widely used refrigerant in vapour

compression refrigeration system (using large centrifugal compressor)?

- a. freon
- b. liquid sulphur dioxide
- c. methyl chloride
- d. ammonia

Ans: A

- 6. The most important application of distribution law is in
  - a. Evaporation
  - b. Liquid extraction
  - c. Drying
  - d. Distillation
    - Ans: B
- 7. Pick out the correct statement.
  - a. a real gas on expansion in vacuum gets heated up
  - b. an ideal gas on expansion in vacuum gets cooled
  - c. an ideal gas on expansion in vacuum gets heated up
  - a real gas on expansion in vacuum cools down whereas ideal gas remains unaffected
    Ans: D

Ans: D

- 8. After throttling, gas temperature
  - a. decreases
  - b. increases
  - c. remains same
  - d. may increase or decrease; depends on the nature of the gas Ans: A
- 9. Which one is true for a throttling process?
  - a. A gas may have more than one inversion temperature
  - b. The inversion temperature is different for different gases
  - c. The inversion temperature is same for all gases
  - d. The inversion temperature is the temperature at which Joule
    - Thomson Co-efficient is infinity
    - Ans: B
- 10. A gas shows deviation from ideal behaviour at
  - a. low pressure and high temperature
  - b. low pressure and low temperature
  - c. low temperature and high pressure
  - d. high temperature and high pressure Ans: C