Chemical Engineering Thermodynamics Objective Questions

Question: 1
Free energy change at equilibrium is
(A) indeterminate
(B) positive
(C) negative
(D) zero
Ans: D
zero
Question: 2
The unit of fugacity is the same as that of the
(A) molar concentration
(B) volume
(C) temperature
(D) pressure
Ans: D
pressure
Question: 3
For an ideal gas the slope of the pressure volume curve, at a given point will be
(A) identical for both the processes
(B) of opposite signs

(C) steeper for an isothermal than for an adiabatic process
(D) steeper for an adiabatic than for an isothermal process
Ans: D
steeper for an adiabatic than for an isothermal process
Question: 4
Reduced pressure of a gas is the ratio of its
(A) pseudocritical pressure to pressure
(B) pressure to pseudocritical pressure
(C) critical pressure to pressure
(D) pressure to critical pressure
Ans: D
pressure to critical pressure
Question: 5
Partial molal quantities are important in the study of
(A) a pure component
(B) non-ideal mixtures
(C) ideal gas
(D) ideal solutions
Ans: B
non-ideal mixtures