## Metall

5. Substitutional diffusion becomes possible, if the

urgical Engineering Physics of Metals Quiz Questions		
1.	Wł	nich of the following properties of metals cannot be explained by free electron
	the	eory of metals?
	a.	Thermal conductivity
	b.	Optical properties
	C.	Electrical conductivity
	d.	Specific heat
		Ans: D
2.	Se	elf diffusion
	a.	Is the migration of atom in pure materials
	b.	Occurs in binary metal alloys
	C.	Involves atomic migration along grain boundaries
3.	d. Involves atomic migration along the surface of a phase  Ans: A  B. Energy associated with an electron at absolute zero temperature (which is the	
	ma	ximum value for that electron) is called the energy.
	a.	Ionisation
	b.	Fermi
	C.	Degenerate
	d.	Electron
		Ans: B
4.	W	hich of the following cyrstallises in bcc structure?
	a.	Chromium
		Lead
	C.	Zinc
	d.	Aluminium
		Ans: A

- a. Vacancies are present in a substitutional solid solution
- b. Solid solution temperature is very high
- c. Atoms are significantly different size
- d. Activation energy for vacancy formation is low

Ans: A

- Photoelectric emission is the emission of electrons from a metallic surface, when it is struck by
  - a. X-rays
  - b. Photons
  - c. Electrons
  - d. None of these

Ans: B

- 7. \_\_\_\_\_ of an atom implies, its capability to attract electrons to itself.
- a. Electronegativity
  b. Electropositivity
  c. Oxidation number
  - d. Electronic defect

Ans: A

- 8. Vander wall bond
  - a. Is a primary bond
  - b. Involves the transfer of electrons between the atoms
  - c. Involves sharing of electrons between the atoms
  - d. Is a weak bond

Ans: D

- 9. X-ray photons possess larger energy compared to the visible light photons because of its comparatively higher
  - a. Wavelengths
  - b. Frequency
  - c. Velocity

d. Plank's constant

Ans: B

- 10. Electrons are assumed to move in a region of \_\_\_\_\_potential, as per zone theory.
  - a. Periodic
  - b. Very low
  - c. Constant
  - d. Very high

Ans: A

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